

Facial selectivity in the hydroboration of androst-4-enes

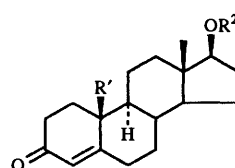
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In the absence of an allylic hydroxy group, the stereochemistry of hydroboration of an androst-4-ene is determined by the presence and stereochemistry of the C-10 methyl group. Allylic hydroxy groups at C-3 direct the hydroboration/oxidation to the *anti*-face. In the case of the 3 α -alcohol, this effect is in opposition to the normal hydration from the α -face of the steroid and leads to the 4 β -alcohol. The stereochemistry of 4 β ,17 β -diacetoxy-19-nor-5 β -androstane was established by X-ray crystallography.

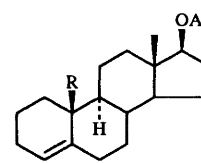
The hydration of alkenes by hydroboration and the subsequent oxidation of the borane with alkaline hydrogen peroxide, is a widely used synthetic reaction.¹ The hydration of an isolated steroidal alkene is reported² to proceed in an anti-Markownikoff sense from the α -face of the molecule. Although 19-nor steroids were not studied, the facial selectivity was attributed to steric hindrance on the β -face of the molecule by the angular methyl group. In smaller molecules an allylic hydroxy group has also been shown to affect the facial selectivity of the hydroboration of an alkene. Thus, the hydroboration of cyclohex-2-enols leads to the *trans*-1,2-diols.^{3,4} The use of hydroboration in the synthesis of steroidal glycols requires an evaluation of the balance between the effect of the 10-methyl group and an allylic hydroxy group in determining the facial selectivity of the addition. In this paper we report on the facial selectivity of the hydroboration of the trisubstituted 4-enes, 17 β -acetoxyandrost-4-ene **5**, 17 β -acetoxy-19-norandrost-4-ene **6** and the retro-steroid, 17 β -acetoxy-9 β ,10 α -androst-4-ene **8** and of the effect of the pseudo-axial and equatorial 3 α - and 3 β -hydroxy groups on the stereochemistry of the reaction. In prior work the hydroboration and oxidation of cholest-4-ene has been reported² to give exclusively 5 α -cholestan-4 α -ol. The hydroboration of testosterone **1** gave 5 α -androstane-3 β ,4 α ,17 β -triol **13**.⁵ The reduction of testosterone acetate under Cagliotti conditions (B₂H₆, Ac₂O) has been examined^{6,7} thoroughly to optimize the conditions for preparing 17 β -acetoxy-5 α -androst-3-ene.

The alkenes were prepared as follows. 17 β -Acetoxyandrost-4-ene **5**, free from isomeric 3-enes, was best obtained⁸ by reduction of testosterone acetate **2** with sodium borohydride in trifluoroacetic acid-acetonitrile-acetic acid-methylene dichloride. 17 β -Acetoxy-19-norandrost-4-ene **6** and the retro-steroid, 17 β -acetoxy-9 β ,10 α -androst-4-ene **8** were prepared in a similar manner from 19-nortestosterone acetate **4** and retrotestosterone acetate **7**, respectively. 17 β -Acetoxyandrost-4-en-3 β -ol **9** was obtained by the reduction of testosterone acetate **2** with sodium borohydride† in methanol. 19-Norandrost-4-ene-3 β ,17 β -diol **10** was obtained in a similar manner from 19-nortestosterone **3**. Inversion of the 3 β -hydroxy group to form 17 β -acetoxyandrost-4-en-3 α -ol **11** using the normal Mitsunobu procedure¹⁰ was accompanied by elimination to form a mixture of dienes. The best yields were obtained by using triphenylphosphine-chloroacetic acid and ethyl azodicarboxylate.¹¹ Selective hydrolysis of the resultant 3 α -chloroacetate under mild conditions with aqueous methanolic potassium carbonate then gave 17 β -acetoxyandrost-4-en-3 α -ol **11**.

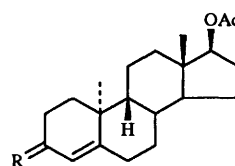
The hydration of the steroids was carried out by



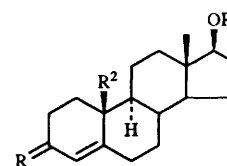
- 1** R¹ = Me, R² = H
2 R¹ = Me, R² = Ac
3 R¹ = R² = H
4 R¹ = H, R² = Ac



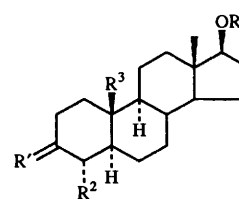
- 5** R = Me
6 R = H



- 7** R = O.
8 R = H₂.



- 9** R¹ = α -H, β -OH; R² = Me; R³ = Ac
10 R¹ = α -H, β -OH; R² = R³ = H
11 R¹ = α -H, β -H; R² = Me; R³ = Ac



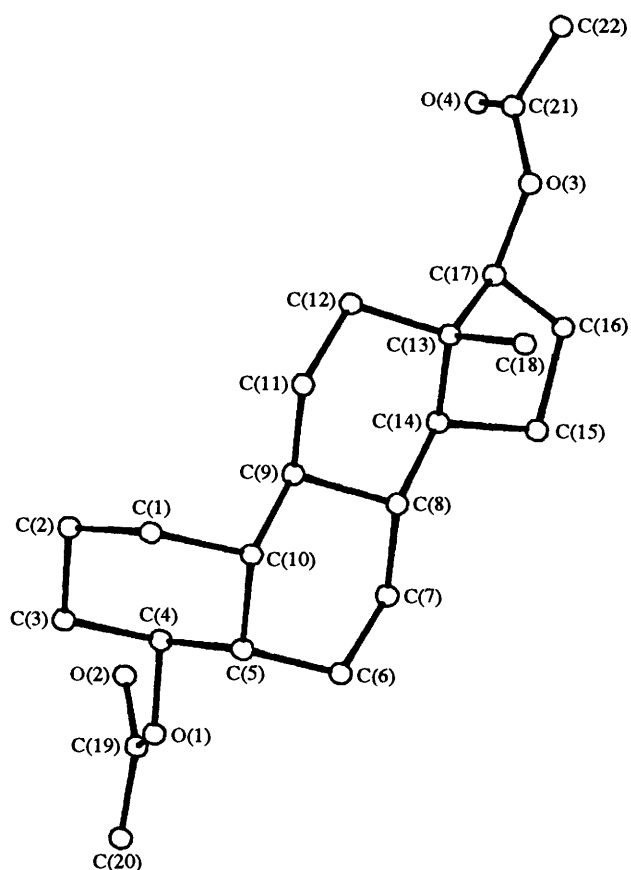
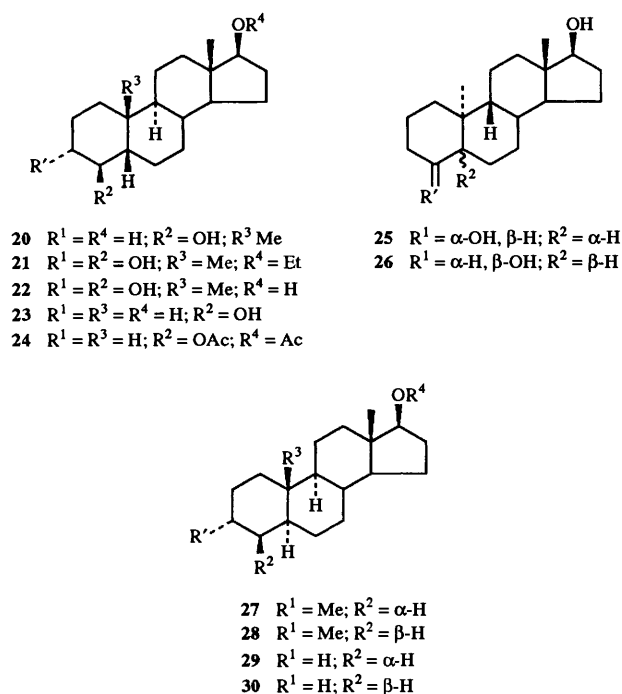
- 12** R¹ = H₂; R² = OH; R³ = Me; R⁴ = H
13 R¹ = α -H, β -OH; R² = OH; R³ = Me; R⁴ = H
14 R¹ = α -H, β -OAc; R² = OAc; R³ = Me; R⁴ = Ac
15 R¹ = α -OH, β -H; R² = R⁴ = H; R³ = Me
16 R¹ = α -H, β -OH; R² = R⁴ = H; R³ = Me
17 R¹ = H₂, R² = OH; R³ = R⁴ = H
18 R¹ = α -H, β -OH; R² = OH; R³ = R⁴ = H
19 R¹ = α -H, β -OAc; R² = OAc; R³ = H; R⁴ = Ac

hydroboration with borane in tetrahydrofuran followed by oxidation with alkaline hydrogen peroxide. The results are given in Table 1. The structures of the products were established as follows. In contrast to the previous work with cholest-4-ene,² the hydroboration and oxidation of 17 β -acetoxyandrost-4-ene **5** gave both 5 α -androstane-4 α ,17 β -diol **12** and 5 β -androstane-4 β ,17 β -diol **20**. Although the 4-H ¹H NMR resonances in each showed a similar multiplicity [δ_{H} 3.41, dt, *J* 4.6 and 10.8 Hz in **12**; δ_{H} 3.56, dt, *J* 4.6 and 10.8 Hz in **20**] arising from two diaxial and one axial:equatorial coupling, the epimers were distinguished by a nuclear Overhauser enhancement of 5% of the

† Now more correctly named as sodium boranuide according to IUPAC rules of nomenclature. Editor.

Table 1 Hydroboration/oxidation of androst-4-enes

Substrate	Product	Yield (%)
17 β -Acetoxyandrost-4-ene	5	
5 α -Androstane-4 α ,17 β -diol	12	81
5 β -Androstane-4 β ,17 β -diol	20	17
17 β -Acetoxy-19-norandrost-4-ene	6	
19-Nor-5 α -androstane-4 α ,17 β -diol	17	35
19-Nor-5 β -androstane-4 β ,17 β -diol	23	38
17 β -Acetoxy-9 β ,10 α -androst-4-ene	8	
5 α ,9 β ,10 α -Androstane-4 α ,17 β -diol	25	14
5 β ,9 β ,10 α -Androstane-4 β ,17 β -diol	26	77
17 β -Acetoxyandrost-4-en-3 β -ol	9	
5 α -Androstane-3 β ,4 α ,17 β -triol	13	
isolated as the triacetate	14	65
19-Norandrost-4-ene-3 β ,17 β -diol	10	
19-Nor-5 α -androstane-3 β ,4 α ,17 β -triol	18	
isolated as the triacetate	19	48
17 β -Acetoxyandrost-4-en-3 α -ol	11	
5 α -Androstane-3 α ,17 β -diol	15	as a 1:2 mixture
5 α -Androstane-3 β ,17 β -diol	16	
17 β -Ethoxy-5 β -androstane-3 α ,4 β -diol	21	21
5 β -Androstane-3 α ,4 β ,17 β -triol	22	63

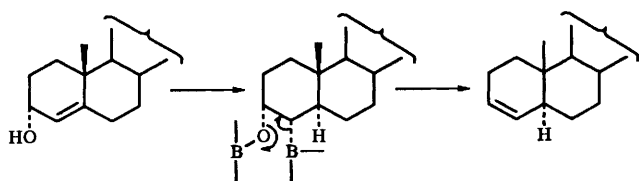
**Fig. 1** X-Ray structure of 4 β ,17 β -diacetoxy-19-nor-5 β -androstane **24**

4 β -H signal (δ_{H} 3.41) in **12** on irradiation of the 19-H resonance. No effect was observed in **20**. The same method was used to distinguish between the isomers **25** and **26** obtained from the retro-steroid **8**. Irradiation of the 19-H resonance at δ_{H} 1.09 in **26** produced an NOE, enhancement of 5% of the CH(OH) resonance at δ_{H} 3.95. However, in the 19-nor series in which the epimers **17** and **23** were obtained in approximately equal amounts, there was no convenient NMR distinction and consequently the stereochemistry of the alcohols was established by X-ray crystallography. Suitable crystals were obtained from 4 β ,17 β -diacetoxy-19-nor-5 β -androstane **24** (see Fig. 1).

In order to confirm their stereochemistry at C-5 both 5 α -androstane-4 α ,17 β -diol **12** and 5 β -androstane-4 β ,17 β -diol **20** were separately oxidized to their corresponding diketones (**27** and **28**). These may be distinguished by the position of their 19-H proton resonances (δ_{H} 0.87 and 1.14, respectively). Equilibration of each with methanolic potassium hydroxide gave the same 3:1 mixture of 5 α - and 5 β -androstane-4,17-

diones with *trans*- and *cis*-fused A/B ring junctions, respectively. The corresponding diketones, **29** and **30**, were also prepared in the 19-nor series. The *cis* A/B fused isomer **30** was converted by base into the *trans* fused 19-nor-5 α -androstane-4,17-dione **29**.

The stereochemistry of the products obtained from the C-3 alcohols was evident from the multiplicity of their CH(OH) ¹H NMR signals. Thus, the 4-H signal in **14** (the triol **13** was purified as its acetate) appeared as a double doublet (*J* 9.5 and 11.5 Hz), in which there was a large coupling (11.5 Hz) to the C-3 α proton corresponding to a diaxial relationship. A similar



Scheme 1

situation was present in **18**. In the case of compound **21** decoupling of the proton resonance at δ_{H} 3.70 (dd, J 9.0 and 9.1 Hz, 4-H) collapsed the signal at δ_{H} 3.39 (ddd, J 5.8, 9.0 and 11.4 Hz, 3-H) to a double doublet (J 5.8 and 11.4 Hz). There is thus a diaxial relationship between these protons and hence a diequatorial relationship between the hydroxy groups. There was a similar coupling pattern in **22**.

A number of conclusions may be drawn from these results. Comparison of the results of hydroboration of 17 β -acetoxyandrost-4-ene **5**, 17 β -acetoxy-19-norandrost-4-ene **6** and 17 β -acetoxy-9 β ,10 α -androst-4-ene **8** reveals the stereochemical directing effect of the 10-methyl group. However, the allylic C-3 hydroxy groups direct hydroboration to the face of the alkene opposite to the hydroxy group. Although 17 β -acetoxyandrost-4-ene and the 19-noralkene both gave some 4 β -alcohol, a 3 β -hydroxy group directed attack entirely to the α -face. On the other hand a 3 α -hydroxy group directed the hydroboration entirely to the β -face despite the steric hindrance from the 10 β -methyl group. The origin of this facial selectivity may be electronic or it may lie in the formation of bulky borate complexes from the alcohol which direct a second molecule of the borane to attack from the *trans* face. The formation of the 3 α ,17 β - and 3 β ,17 β -dihydroxy-5 α -androstanes (**15** and **16**) may occur through an elimination reaction of the borane to form a 3-ene (see Scheme 1) which is then rehydroborated. The ethyl ether in **21** presumably arose by reduction of the acetate.

Experimental

Light petroleum refers to the fraction bp 60–80 °C; silica for chromatography was Merck Kieselgel 60 type 9380. IR spectra were determined as Nujol mulls; ^1H NMR spectra were determined at 360 or 500 MHz in deuteriochloroform on Bruker WM 360 or AMX spectrometers; J values given in Hz. Borane (1 mol dm $^{-3}$) in tetrahydrofuran was used as supplied by Aldrich Chemicals. Extracts were dried over anhydrous sodium sulfate.

Preparation of 17 β -acetoxyandrost-4-ene **5**⁸

Sodium borohydride (1.3 g) was added in portions with stirring and cooling to a previously cooled mixture of trifluoroacetic acid (8.0 cm 3), acetic acid (8.0 cm 3) and acetonitrile (8.0 cm 3). 17 β -Acetoxyandrost-4-en-3-one (2.3 g)(**2**) in dry dichloromethane (40 cm 3) was added to the mixture which was then stirred at room temperature for 1 h. After this the reaction mixture was neutralized with saturated aqueous sodium hydrogen carbonate and extracted with dichloromethane. The combined extracts were washed with water, dried and evaporated under reduced pressure to give 17 β -acetoxyandrost-4-ene **5** (2.2 g), which crystallized from light petroleum as prisms, mp 89 °C (lit.,⁸ 97 °C) $\nu_{\text{max}}/\text{cm}^{-1}$ 1732 and 1625; δ_{H} 0.80 (3 H, s, 18-H), 1.05 (3 H, s, 19-H), 2.10 (3 H, s), 4.56 (1 H, t, J 8.4, 17-H) and 5.31 (1 H, t, J 3.5, 4-H). 17 β -Acetoxy-19-norandrost-4-ene **6**, mp 79–81 °C (lit.,¹³ 81–82 °C) and 17 β -acetoxy-9 β ,10 α -androst-4-ene **8** were prepared in a similar manner. The latter crystallized from light petroleum as prisms, mp 51–53 °C (Found: C, 79.2; H, 10.0. C $_{21}$ H $_{32}$ O $_2$ requires C, 79.7; H, 10.2%); $\nu_{\text{max}}/\text{cm}^{-1}$ 1736 and 1660; δ_{H} 0.82 (3 H, s, 18-H) 1.14 (3 H, s, 19-H), 3.53 (1 H, t, J 8.4, 17-H) and 5.28 (1 H, br s, 4-H).

Preparation of 19-norandrost-4-ene-3 β ,17 β -diol **10**

17 β -Hydroxy-19-norandrost-4-en-3-one **3** (2 g) in methanol (40 cm 3) was treated with sodium borohydride (500 mg) at 0 °C for 2 h. Acetic acid (0.4 cm 3) was added to the mixture which was then concentrated under reduced pressure. Water was added to the residue and the products were recovered in ethyl acetate. The extract was washed with aqueous sodium hydrogen carbonate and brine, dried and evaporated to give 19-norandrost-4-ene-3 β ,17 β -diol **10** (1.87 g) which crystallized from ethyl acetate–light petroleum as needles, mp 143 °C (Found: C, 77.6; H, 10.4. C $_{18}$ H $_{28}$ O $_2$ requires C, 78.2; H, 10.2%); $\nu_{\text{max}}/\text{cm}^{-1}$ 3350; δ_{H} 0.77 (3 H, s, 18-H), 3.63 (1 H, t, J 8, 17-H), 4.14 (1 H, m, 3-H) and 5.38 (1 H, s, 4-H).

17 β -Acetoxyandrost-4-en-3 β -ol **9**, prepared from 17 β -acetoxyandrost-4-en-3-one **2** under similar conditions crystallized from light petroleum as needles, mp 135–136 °C (lit.,⁹ 135–136 °C) $\nu_{\text{max}}/\text{cm}^{-1}$ 3310 and 1735; δ_{H} 0.83 (3 H, s, 18-H), 1.10 (3 H, s, 19-H), 2.03 (3 H, s, 17-OAc), 4.13 (1 H, m, 3-H), 4.60 (1 H, t, J 8.4, 17-H) and 5.31 (1 H, br s, 4-H).

Preparation of 17 β -acetoxy-3 α -chloroacetoxyandrost-4-ene

Triphenylphosphine (3.0 g) and dry chloroacetic acid (1.2 g) were added to a solution of 17 β -acetoxyandrost-4-en-3 β -ol (2.0 g) in toluene (60 cm 3). Diethyl azodicarboxylate (1 cm 3) was then added dropwise to the mixture after which it was stirred overnight at room temperature. After evaporation of the solvent under reduced pressure from the mixture the residue was chromatographed on silica. Elution with 20% ethyl acetate–light petroleum gave the *title compound* (1.4 g) which crystallized from ethyl acetate–light petroleum as needles, mp 128–130 °C (Found: C, 67.0; H, 7.7. C $_{23}$ H $_{33}$ ClO $_4$ requires C, 67.5; H, 8.1%); $\nu_{\text{max}}/\text{cm}^{-1}$ 1760, 1737 and 1651; δ_{H} 0.82 (3 H, s, 18-H), 1.01 (3 H, s, 19-H), 2.04 (3 H, s, OAc), 4.05 (2 H, s, OCOCH $_2$ Cl), 4.58 (1 H, br s, 3 β -H) and 5.45 (1 H, d, J 5, 4-H).

Preparation of 17 β -acetoxyandrost-4-en-3 α -ol **11**

An aqueous solution (10 cm 3) of potassium carbonate (2.0 g) was added to a solution of 17 β -acetoxy-3 α -chloroacetoxyandrost-4-ene (1.0 g) in methanol (30 cm 3) and the mixture kept at room temperature for 1 h. After this acetic acid (1 cm 3) was added to the mixture which was then concentrated under reduced pressure and the product recovered in ethyl acetate. The extract was washed with aqueous sodium hydrogen carbonate and water, dried and evaporated. The residue was chromatographed on silica, eluting with 10% ethyl acetate–light petroleum, to give the *title compound* **11** (650 mg) which crystallized from ethyl acetate–light petroleum as needles, mp 108–110 °C (Found: C, 74.6; H, 9.4. C $_{21}$ H $_{32}$ O $_3$ ·0.5H $_2$ O requires C, 73.9; H, 9.7%); $\nu_{\text{max}}/\text{cm}^{-1}$ 3230, 1730 and 1660; δ_{H} 0.81 (3 H, s, 18-H), 0.99 (3 H, s, 19-H), 2.04 (3 H, s, OAc), 4.07 (1 H, br s, 3 β -H), 4.56 (1 H, t, J 8.7, 17 α -H) and 5.47 (1 H, s, 4-H).

Hydroboration/oxidation experiments

(a) 17 β -Acetoxyandrost-4-ene **5** (1.0 g) in dry tetrahydrofuran (30 cm 3) was treated with borane in tetrahydrofuran (1 mol dm $^{-3}$; 20 cm 3) at 0 °C under nitrogen. The reaction was followed by TLC until the formation of the borane was complete (4 h). The mixture was carefully diluted with water (10 cm 3) after which it was cooled to 0 °C and treated with sodium hydroxide (10%; 20 cm 3) followed by hydrogen peroxide (30%; 30 cm 3), added dropwise. The mixture was stirred overnight after which sodium sulfite (2 g) was added to it followed by acetic acid (1 cm 3), water (50 cm 3), dil. hydrochloric acid (50 cm 3) and ethyl acetate (100 cm 3). The mixture was stirred for a further 15 min after which the organic layer was separated, washed with water and brine, dried and evaporated under reduced pressure to give a residue. This was chromatographed on silica, eluting with 20% ethyl acetate–light petroleum to give

5 α -androstane-4 α ,17 β -diol **12** (746 mg) which crystallized from ethyl acetate–light petroleum as plates, mp 232–234 °C (lit.,¹² mp 235–237 °C); $\nu_{\max}/\text{cm}^{-1}$ 3600 and 3453; δ_{H} 0.72 (3 H, s, 18-H), 0.80 (3 H, s, 19-H), 3.41 (1 H, dt, *J* 4.6 and 10.8, 4 β -H) and 3.59 (1 H, t, *J* 8.6, 17 α -H). Further elution with the same solvent system gave 5 β -androstane-4 β ,17 β -diol **20** (162 mg) which crystallized from acetone as prisms, mp 178–181 °C (lit.,¹² mp 177–178 °C); $\nu_{\max}/\text{cm}^{-1}$ 3585 and 3448; δ_{H} 0.71 (3 H, s, 18-H), 0.99 (3 H, s, 19-H), 3.56 (1 H, dt, *J* 4.6 and 10.8, 4 α -H) and 3.85 (1 H, t, *J* 8.6, 17 α -H).

(b) 17 β -Acetoxyandrost-4-ene-3 β -ol **9** (2 g) in dry tetrahydrofuran (50 cm³) was treated with borane in tetrahydrofuran (1 mol dm⁻³; 40 cm³) at 0 °C under nitrogen for 4 h (TLC control) after which the mixture was carefully diluted with water (10 cm³) and cooled to 0 °C. Sodium hydroxide (10%; 20 cm³) was then added to the mixture followed by hydrogen peroxide (30%; 40 cm³), added dropwise. The mixture was stirred overnight after which sodium sulfite (2 g) added followed by acetic acid (1 cm³), water (50 cm³), dil. hydrochloric acid (50 cm³) and ethyl acetate (100 cm³) were added to it. The mixture was stirred for a further 15 min after which the organic layer was separated, washed with water and brine, dried and evaporated under reduced pressure. The residue (1.93 g) was treated with acetic anhydride (4 cm³) in pyridine (10 cm³) overnight and the mixture worked up to provide a product which was chromatographed on silica. Elution with 20% ethyl acetate–light petroleum gave 3 β ,4 α ,17 β -triacetoxy-5 α -androstane **14** (1.70 g) which crystallized from ethyl acetate as needles, mp 124–126 °C (Found: C, 69.0; H, 9.1. C₂₅H₃₈O₆ requires C, 69.1; H, 8.8%); $\nu_{\max}/\text{cm}^{-1}$ 1750 and 1730; δ_{H} 0.78 (3 H, s, 18-H), 0.92 (3 H, s, 19-H), 2.00, 2.03 and 2.04 (each 3 H, s, OAc), 4.58 (1 H, t, *J* 8.4, 17 α -H), 4.76 (1 H, ddd, *J* 5.5, 9.4, and 11.5, 3 α -H) and 4.97 (1 H, dd, *J* 9.5 and 11.5, 4 β -H).

(c) 17 β -Acetoxyandrost-4-ene-3 α -ol **11** (600 mg) in dry tetrahydrofuran (25 cm³) was treated with borane in tetrahydrofuran, (1 mol dm⁻³; 25 cm³) at 0 °C under nitrogen for 4 h (TLC control) after which the mixture was diluted with water (10 cm³) and cooled to 0 °C. Sodium hydroxide (10%; 15 cm³) was then added to the mixture followed by hydrogen peroxide (30%; 20 cm³) added dropwise. The mixture was stirred overnight after which sodium sulfite (2 g) was added to it followed by acetic acid (1 cm³), water (50 cm³), dil. hydrochloric acid (50 cm³) and ethyl acetate (100 cm³). The mixture was stirred for a further 15 min after which the organic layer was separated, washed with water and brine, dried and evaporated. The residue was chromatographed on silica, eluting with 15% ethyl acetate–light petroleum to give a mixture (55 mg) of 5 α -androstane-3 α ,17 β - and 3 β ,17 β -diols **15** and **16**, in the ratio 3 : 2. They were identified by comparison of the ¹H NMR spectrum with those of authentic samples. Further elution with 25% ethyl acetate–light petroleum gave 17 β -ethoxy-5 β -androstane-3 α ,4 β -diol **21** (125 mg), mp 209–212 °C (Found: C, 75.0; H, 10.8. C₂₁H₃₆O₃ requires C, 74.95; H, 10.7%); $\nu_{\max}/\text{cm}^{-1}$ 3583 and 3431; δ_{H} 0.74 (3 H, s, 18-H), 0.97 (3 H, s, 19-H), 1.17 (3 H, t, *J* 7.0, OCH₂CH₃), 3.29 (1 H, t, *J* 8.5, 17-H), 3.39 (1 H, ddd, *J* 5.8, 9.0 and 11.4, 3-H), 3.50 (2 H, m, OCH₂CH₃) and 3.70 (1 H, dd, *J* 9.0, 9.1, 4-H). Further elution with the same solvent system gave 5 β -androstane-3 α ,4 β ,17 β -triol **22** (345 mg), mp 225–227 °C (Found: C, 70.0; H, 10.3. C₁₉H₃₂O₃·H₂O requires C, 69.9; H, 10.5%); $\nu_{\max}/\text{cm}^{-1}$ 3470 and 3338; δ_{H} 0.73 (3 H, s, 18-H), 0.99 (3 H, s, 19-H), 3.60 (1 H, t, *J* 8.5, 17-H), 3.45 (1 H, ddd, *J* 5.8, 8.9 and 11.4, 3 β -H), 3.66 (1 H, dd, *J* 8.9 and 9.0, 4 α -H).

(d) 17 β -Acetoxy-19-norandrost-4-ene **6** (1 g) was treated with borane in tetrahydrofuran as above. The product was chromatographed on silica, eluting with 20% ethyl acetate–light petroleum to give 19-nor-5 α -androstane-4 α ,17 β -diol **17** (318 mg) which crystallized from ethyl acetate–light petroleum as needles, mp 209–212 °C (Found: C, 77.1; H, 11.1. C₁₈H₃₀O₂

requires C, 77.6; H, 10.9%); $\nu_{\max}/\text{cm}^{-1}$ 3415; δ_{H} 0.72 (3 H, s, 18-H), 3.20 (1 H, dt, *J* 4.3, 9.9, 4 β -H), 3.66 (1 H, t, *J* 8.6, 17 α -H). Further elution gave 19-nor-5 β -androstane-4 β ,17 β -diol **23** (349 mg) as prisms, mp 159–161 °C (Found: C, 77.3; H, 10.9. C₁₈H₃₀O₂ requires C, 77.6; H, 10.9%); $\nu_{\max}/\text{cm}^{-1}$ 3300; δ_{H} 0.75 (3 H, s, 18-H), 3.65 (1 H, dt, *J* 4.1 and 11.0, 4-H) and 3.82 (1 H, t, *J* 8.5, 17-H). The diacetate, prepared with acetic anhydride in pyridine, had mp 109–110 °C (Found: C, 83.7; H, 10.9. C₂₂H₃₄O₄ requires C, 84.0; H, 10.9%); $\nu_{\max}/\text{cm}^{-1}$ 1740; δ_{H} 0.78 (3 H, s, 18-H), 2.03 (6 H, s, OAc), 4.59 (1 H, dd, *J* 7.9 and 9.0, 17-H) and 5.10 (1 H, dt, *J* 5.0 and 11.1, 4-H).

(e) 19-Norandrost-4-ene-3 β ,17 β -diol **10** (1.5 g) was treated with borane in tetrahydrofuran as above and the product (1.32 g) was treated with acetic anhydride (3 cm³) in pyridine (7 cm³) at room temperature overnight. The solution was poured into dil. hydrochloric acid and the steroids were recovered in ethyl acetate and chromatographed on silica. Elution with 15% ethyl acetate–light petroleum gave 3 β ,4 α ,17 β -triacetoxy-19-nor-5 α -androstane **19** (1.1 g) which crystallized from ethyl acetate–light petroleum as plates, mp 233 °C (Found: C, 68.5; H, 8.6. C₂₄H₃₆O₆ requires C, 68.6; H, 8.5%); $\nu_{\max}/\text{cm}^{-1}$ 1737; δ_{H} 0.79 (3 H, s, 18-H), 2.00 (3 H, s), 2.02 (6 H, s), (3 × OAc), 4.59 (1 H, t, *J* 8, 17 α -H) and 4.75 (2 H, m, 3- and 4-H). When the spectrum was determined in C₆D₆ this signal was resolved to δ 4.93 (1 H, t, *J* 10, 4-H), 5.02 (1 H, ddd, *J* 4.9, 10 and 11.5, 3-H).

(f) 17 β -Acetoxy-9 β ,10 α -androst-4-ene **8** (1.0 g) was treated with borane in tetrahydrofuran as above. The product was chromatographed on silica, eluting with 20% ethyl acetate–light petroleum to give 5 α ,9 β ,10 α -androstane-4 α ,17 β -diol **25** (130 mg), mp 151–153 °C (Found: C, 77.7; H, 10.9. C₁₉H₃₂O₂ requires C, 78.0; H, 11.0%); $\nu_{\max}/\text{cm}^{-1}$ 3520 and 3453; δ_{H} 0.76 (3 H, s, 18-H), 0.92 (3 H, s, 19-H), 3.48 (1 H, dt, *J* 4.5 and 11.0, 4-H) and 3.68 (1 H, t, *J* 8.6, 17-H). Further elution gave 5 β ,9 β ,10 α -androstane-4 β ,17 β -diol **26** (709 mg) mp 181–183 °C (Found: C, 77.6; H, 10.8. C₁₉H₃₂O₂ requires C, 78.0; H, 11.0%); $\nu_{\max}/\text{cm}^{-1}$ 3613 and 3443; δ_{H} 0.76 (3 H, s, 18-H), 1.09 (3 H, s, 19-H), 3.68 (1 H, t, *J* 8.8, 17-H) and 3.95 (1 H, dt, *J* 4.6 and 10.9, 4-H). Irradiation at δ 1.09 produced an NOE enhancement of 5% at δ_{H} 3.95 m.

Oxidation of 5 α -androstane-4 α ,17 β -diol **12**

The diol **12** (100 mg) in acetone (25 cm³) was treated with Jones' reagent (0.8 cm³) at room temperature for 20 min after which methanol was added to the mixture to destroy the excess of the reagent and the solution was concentrated under reduced pressure. Water was added to the residue and the steroids were recovered in ethyl acetate. The extract was washed with aqueous sodium hydrogen carbonate and water, dried and evaporated to give 5 α -androstane-4,17-dione **27** (80 mg) which crystallized from light petroleum as needles, mp 163–165 °C (lit.,¹³ 162–164 °C); $\nu_{\max}/\text{cm}^{-1}$ 1739 and 1707; δ_{H} 0.77 (3 H, s, 18-H), 0.87 (3 H, s, 19-H). Under similar conditions (i) 5 β -androstane-4 β ,17 β -diol **20** gave 5 β -androstane-4,17-dione **28** which crystallized from light petroleum as needles, mp 158–160 °C (Found: C, 78.85; H, 10.1. C₁₉H₂₈O₂ requires C, 79.1; H, 9.8%); $\nu_{\max}/\text{cm}^{-1}$ 1740 and 1703; δ_{H} 0.85 (3 H, s, 18-H) and 1.14 (3 H, s, 19-H). (ii) 19-Nor-5 α -androstane-4 α ,17 β -diol **17** gave 19-nor-5 α -androstane-4,17-dione **29** which crystallized from ethyl acetate–light petroleum as plates, mp 149–151 °C (Found: C, 78.4; H, 9.1. C₁₈H₂₆O₂ requires C, 78.8; H, 9.5%); $\nu_{\max}/\text{cm}^{-1}$ 1734 and 1704; δ_{H} 0.88 (3 H, s, 18-H). (iii) 19-Nor-5 β -androstane-4 β ,17 β -diol **23** gave 19-nor-5 β -androstane-4,17-dione **30** which crystallized from light petroleum as needles, mp 93–95 °C (Found: C, 78.3; H, 9.1. C₁₈H₂₆O₂ requires C, 78.8; H, 9.5%); $\nu_{\max}/\text{cm}^{-1}$ 1733 and 1705; δ_{H} 0.88 (3 H, s, 18-H).

Equilibration experiments

(i) 5 α -Androstane-4,17-dione **27** (200 mg) in methanol (5 cm³) was treated with aqueous potassium hydroxide [200 mg in

water (0.5 cm³) under reflux for 1 h. The solution was acidified with dil. hydrochloric acid, concentrated under reduced pressure and the steroids recovered in ethyl acetate. The extract was washed with aqueous sodium hydrogen carbonate and brine, dried and evaporated to give a 3:1 mixture of 5 α - and 5 β -androstane-4,17-dione, **27** and **28**, based on the intensity of the 18-H and 19-H ¹H NMR signals at δ_{H} 0.77, 0.87 and 1.14. (ii) Under similar conditions 5 β -androstane-4,17-dione (40 mg) gave the same mixture of **27** and **28**. (iii) 19-Nor-5 β -androstane-4,17-dione **30** (30 mg) under similar conditions gave 19-nor-5 α -androstane-4,17-dione **29** (25 mg), mp 148–150 °C, identified by its IR and ¹H NMR spectra. The 19-nor-5 α -androstane-4,17-dione **29** was recovered unchanged.

X-Ray crystal structure determination of compound **24**

Crystal data. C₂₂H₃₂O₄, *M* 360.5, monoclinic, space group *P*₂₁ (No 4), *a* = 7.481(2), *b* = 10.855(7), *c* = 13.256(9) Å, α = 90°, β = 100.91(4), γ = 90°, *V* = 1057.0 (Å³), *Z* = 2, *D*_c = 1.13 g cm⁻³, *F*(000) = 392, monochromated Mo-K α radiation, λ = 0.710 69 Å, μ = 0.71 cm⁻¹.

Data collection. Data were collected using a crystal, 0.3 × 0.3 × 0.1 mm on an Enraf-Nonius CAD₄ diffractometer. Reflections were measured using a θ -2 θ scan. The data reflection ranges were (θ min. and max.) *h* 0–8, *k* 0–12, *l* –15–15. A total of 2105 reflections were measured of which 1962 were unique reflections and 1070 significant reflections with $|F^2| > 3\sigma(F^2)$ were used in the refinement where $\sigma(F^2) = \{\sigma^2(I) + (0.04 I)^2\}^{1/2}/L_p$. The maximum change in the standard reflections was 0.6%. There was no correction for decay or absorption.

Structure solution and refinement

The structure was solved by direct methods using SHELXS-86. Non-hydrogen atoms were refined anisotropically by full matrix least-squares using the Enraf-Nonius MoLEN program. Hydrogen atoms were held fixed at calculated positions with $U_{\text{iso}} = 1.3 U_{\text{eq}}$ for the parent atom except that the hydrogen atoms attached to C(20) and C(22) (the acetoxy methyl groups) were omitted. The final residuals were *R* = 0.085 and *R'* = 0.106, *S* = 4.07 amu. The number of variables was 234, $(\Delta/\sigma)_{\text{max}}$ 0.02; $(\Delta_p)_{\text{max,min}}$ (e Å⁻³) = +0.28, –0.22. The absolute structure was assigned from the known origin of the material.

The fractional atomic co-ordinates, bond lengths and angles and thermal parameters have been deposited with the Cambridge Crystallographic Data Centre.†

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† For details see Instructions for Authors (1995), *J. Chem. Soc., Perkin Trans. 1*, 1995, Issue 1

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